



Boston

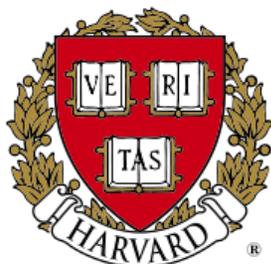
Massachusetts

220<sup>th</sup> ECS Meeting

October 9-14, 2011



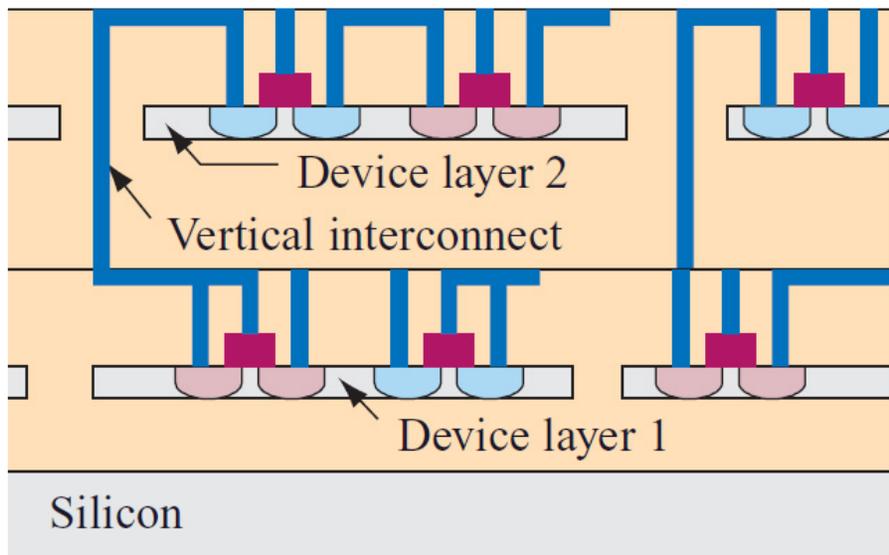
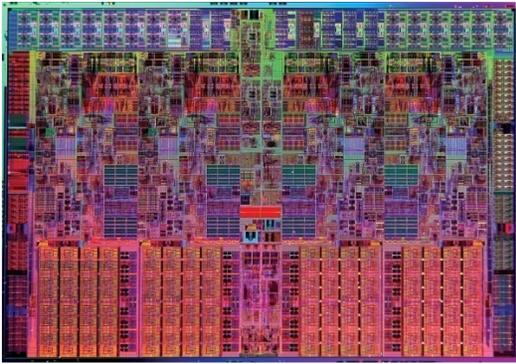
# Vapor Deposition of Highly Conformal Copper Seed Layers for Plating through-Silicon Vias (TSVs)



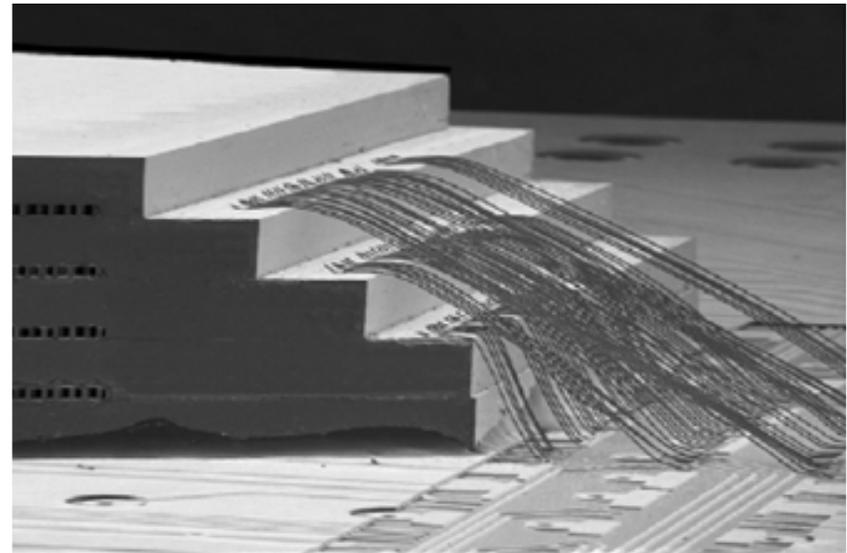
Yeung Au and Roy G. Gordon  
Department of Chemistry and Chemical Biology  
Harvard University

## Introduction

- IC integration has traditionally been done using 2D approaches
- More efficient and compact packages with greater integration flexibility can be achieved by vertical (3D) integration



**3D Integration**



**3D Wire Bonding**

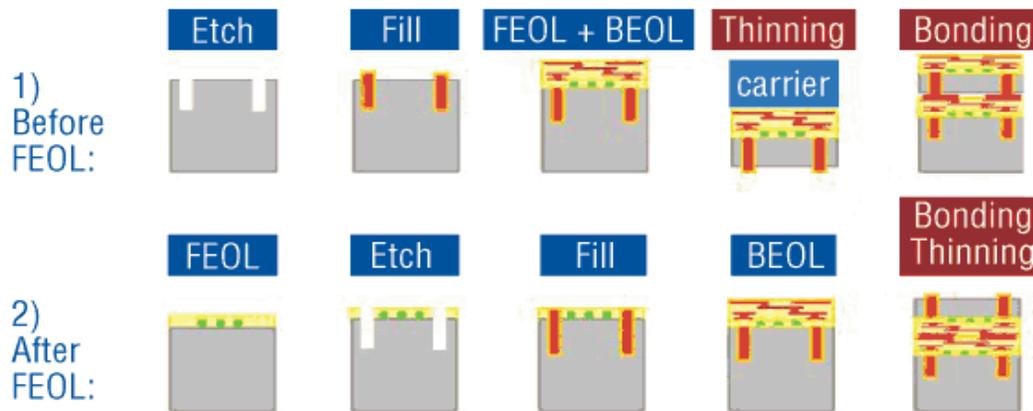
### Limitations of wire bonding

- Long connection
- Excess resistance
- Not scalable

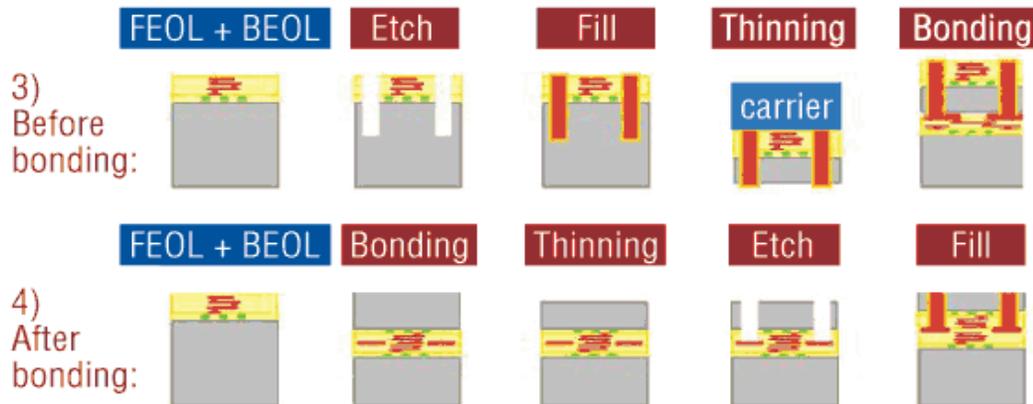
# Introduction

- Through-silicon vias (TSV) enable the formation of higher density and higher aspect ratio connections, allowing better packing density than traditional 3D methods

## “Via-first” approach: before or after FEOL



## “Via-last” approach: before or after bonding/thinning



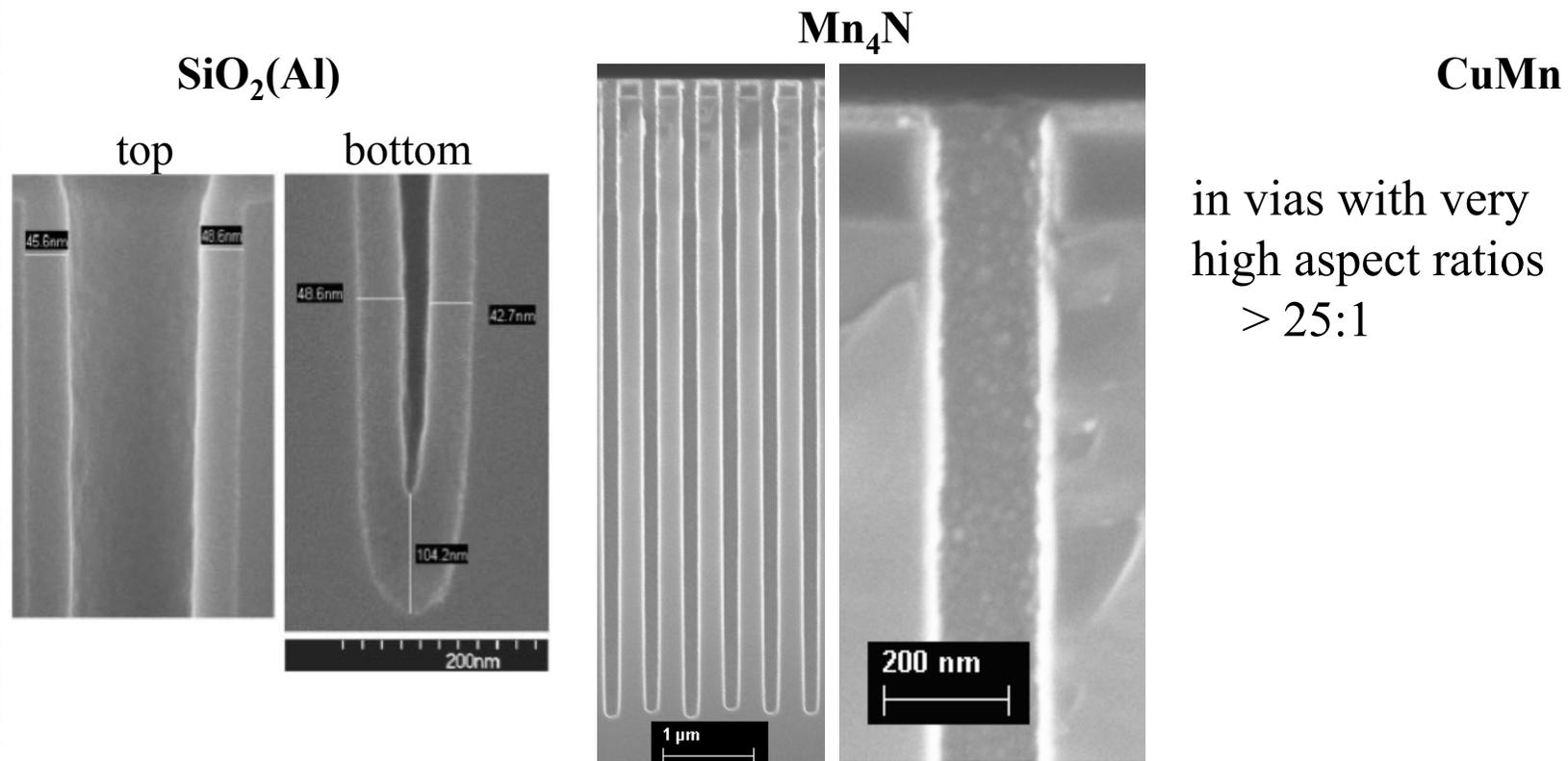
## Main Process Steps

- Via etching in Si
- Insulator deposition
- Barrier deposition**
- Copper seed layer deposition**
- Via filling
- Surface copper removal
- Wafer thinning
- Wafer alignment, bonding, and dicing

- The 2009 ITRS roadmap calls for copper vias with AR of 10:1 by 2012 and 20:1 by 2015

## Introduction

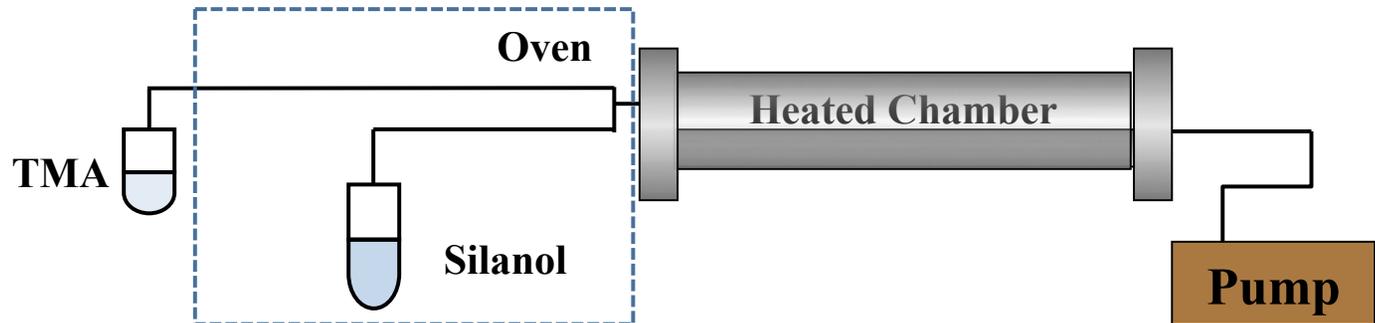
- Rapid atomic layer deposition (ALD) of alumina-doped silica insulating layer
- Chemical vapor deposition (CVD) of manganese nitride barrier/adhesion layer
- ALD of copper seed layer, or
- Direct liquid injection (DLI) CVD of copper manganese alloy seed layer



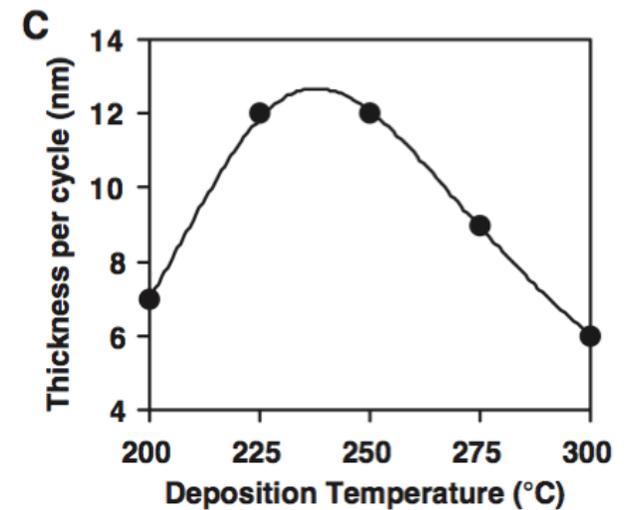
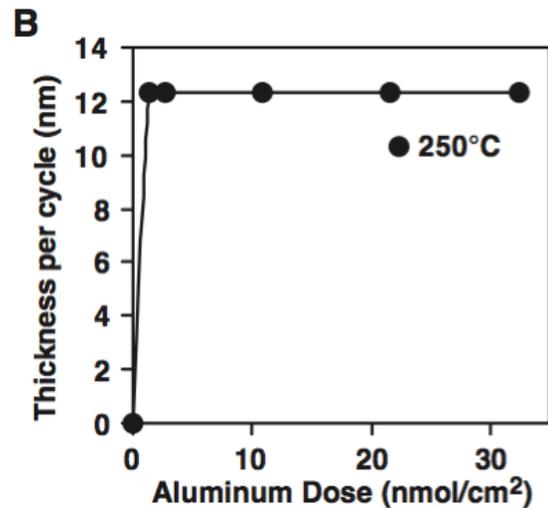
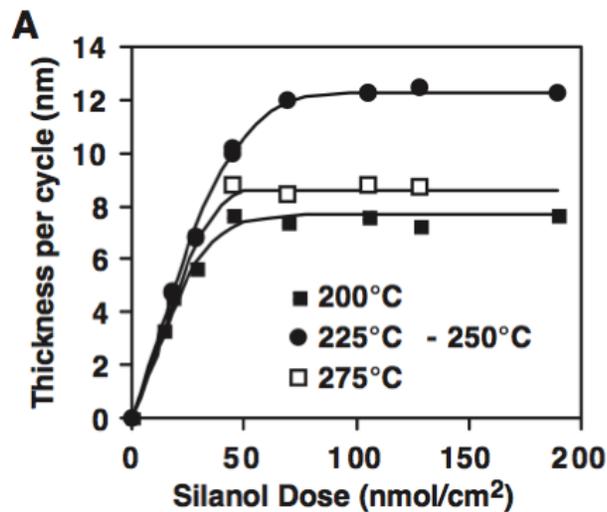
# ALD of Aluminum-doped Silica

- Highly conformal silica film can be deposited using trimethyl aluminum (TMA) and tris(*tert*-butoxy) silanol [(Bu<sup>t</sup>O)<sub>3</sub>SiOH]

## ALD System

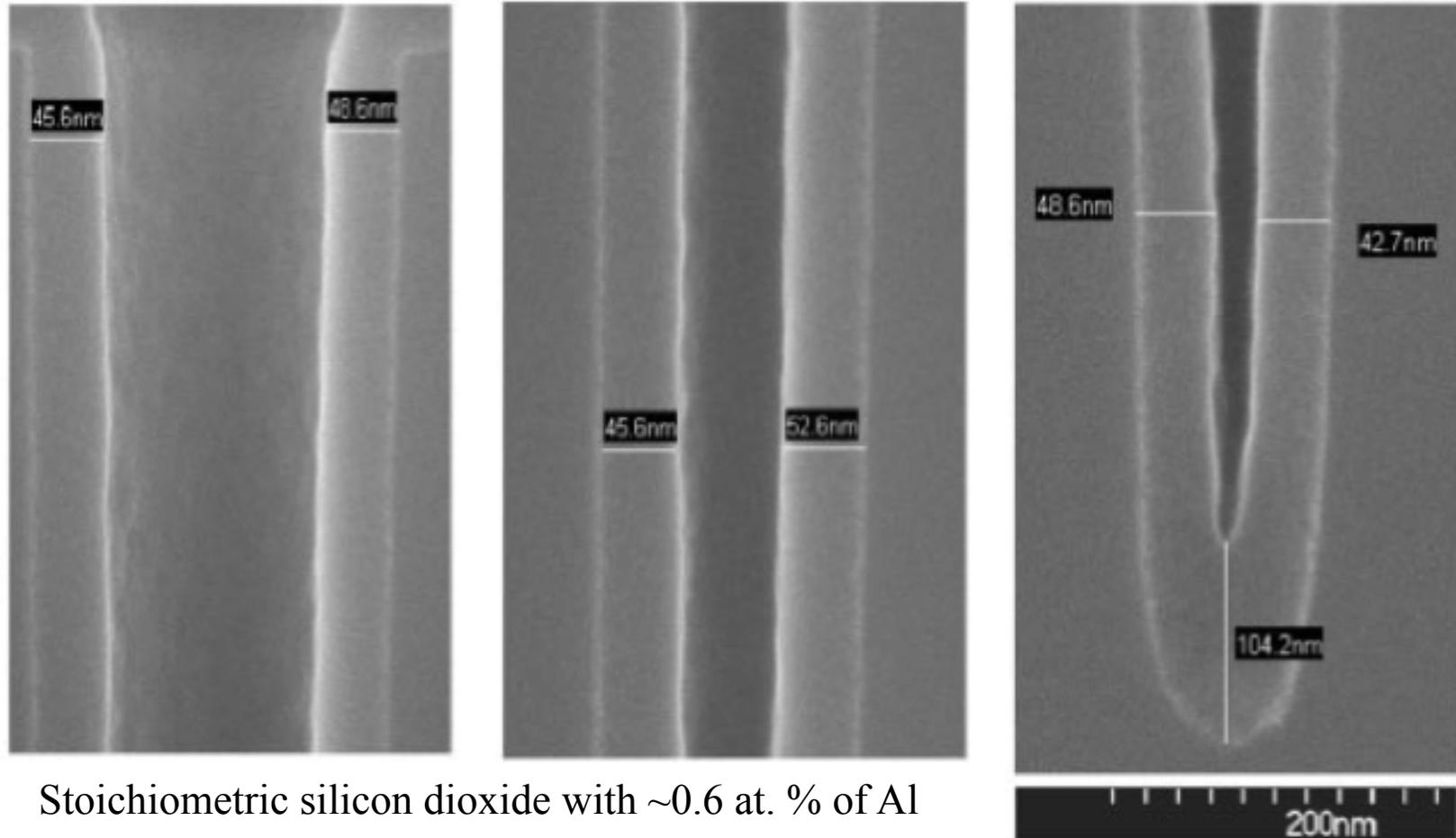


- Saturation behavior and temperature dependence of growth rate:



## ALD of Aluminum-doped Silica

- ALD-silica process allows uniform lining of long narrow holes, aspect ratio  $> 50:1$



- Stoichiometric silicon dioxide with  $\sim 0.6$  at. % of Al
- Amorphous film with density  $2.0 \text{ g/cm}^3$  ( $\sim 91\%$  of bulk silica)

D. Hausmann, J. Becker, S. Wang, and R. G. Gordon, *Science*, **298** 402-406 (2002)

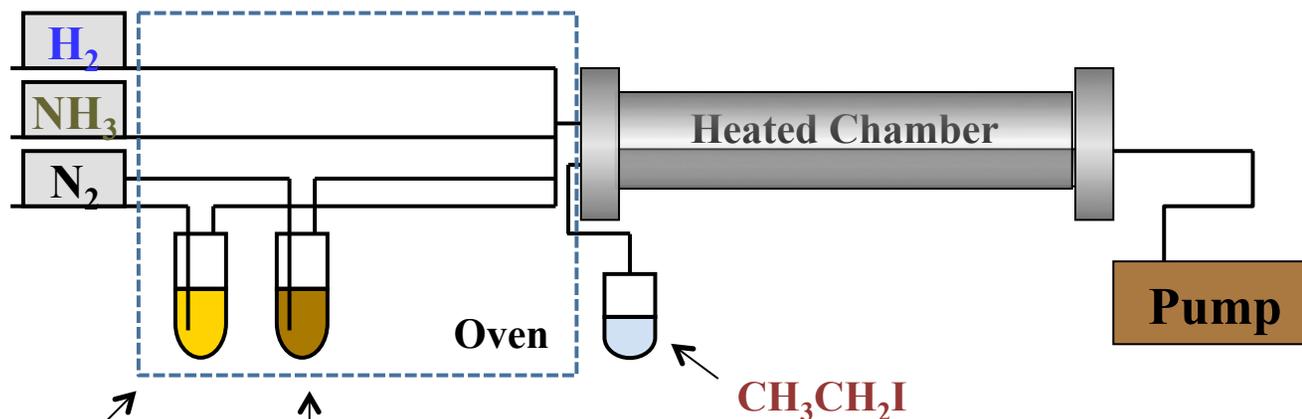
**Highly conformal silica insulating film can be deposited using rapid (up to 12 nm/cycle) ALD in vias with AR over 50:1**

# CVD of Manganese Nitride

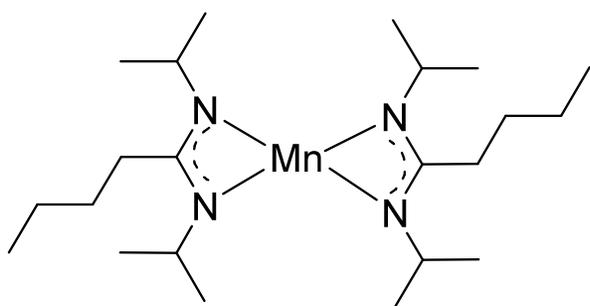
7

## CVD System

Temperature: 130°C  
Pressure: 5 Torr



## Precursors

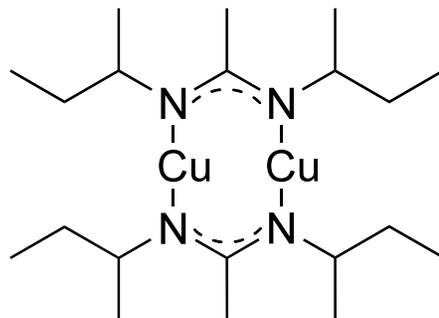


*Bis (N,N'-diisopropylpentylamidinato) manganese(II)*

Melting Point: ~60°C

Bubbler Temperature: 90°C

Vapor Pressure: ~0.1 mbar at 90°C



*Copper (I) N,N'-di-sec-butylacetamidinate*

Melting Point: ~75°C

Bubbler Temperature: 130°C

Vapor Pressure: ~0.25 mbar at 95°C

## Advantages of metal amidinate precursors:

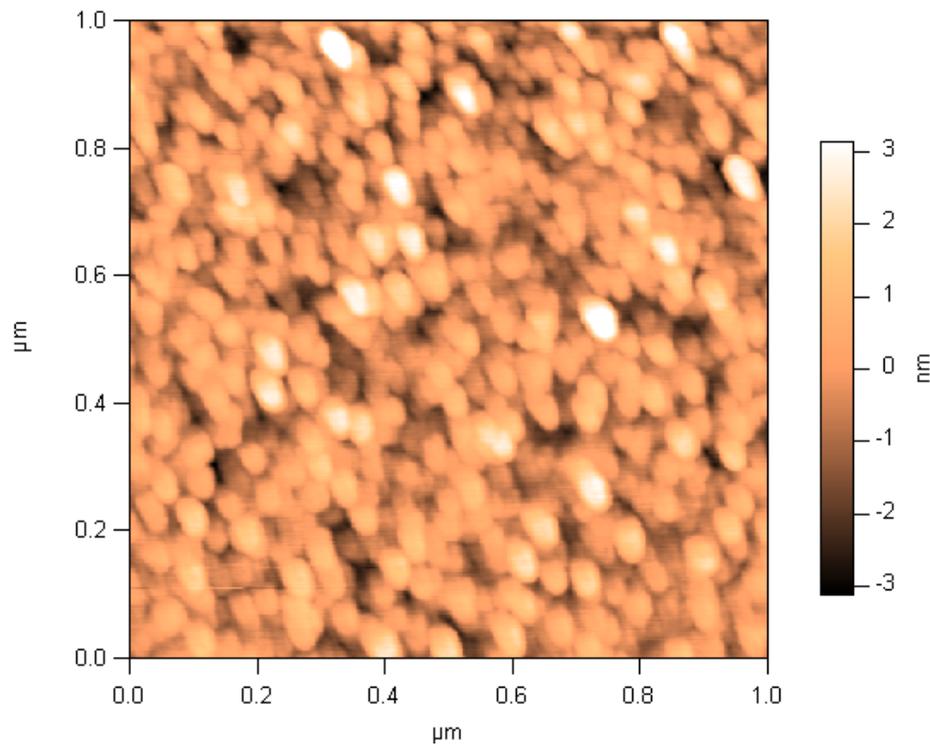
- Chelating effect enhances thermal stability
- Low carbon and oxygen contamination
- Tunable reactivity, volatility, and melting point

B. S. Lim, A. Rahtu, J. S. Park, and R. G. Gordon, *Inorg. Chem.*, **42 (24)**, 7951-7958, (2003).

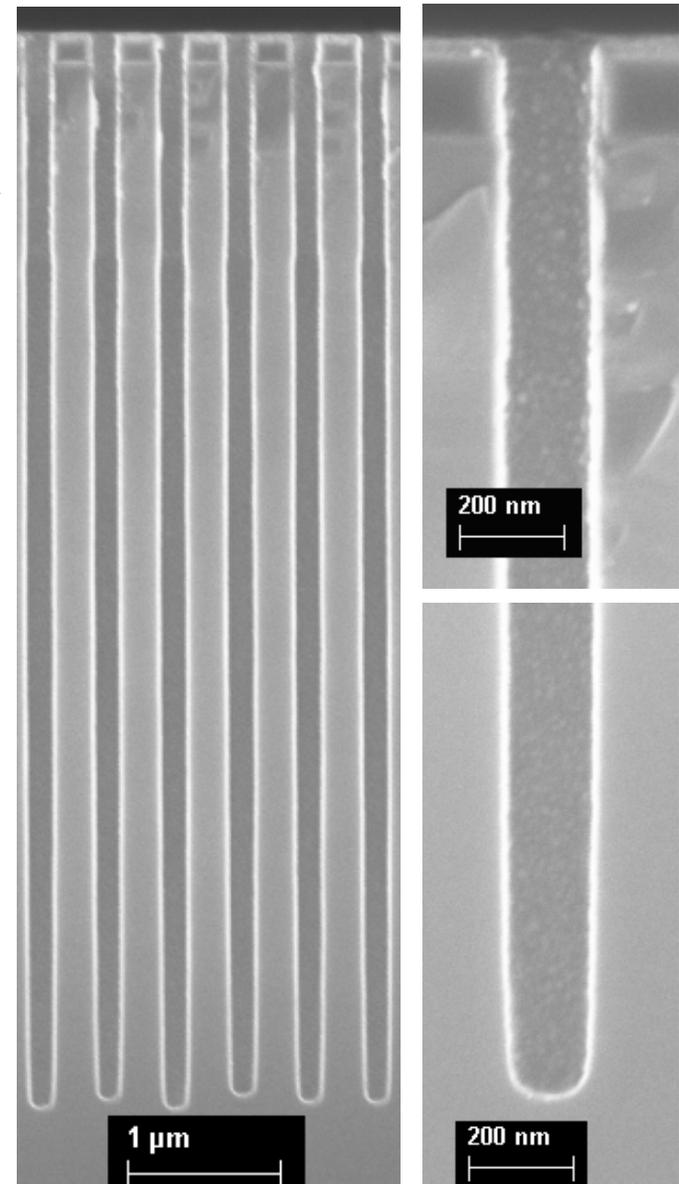
# CVD of Manganese Nitride

- CVD-Mn<sub>4</sub>N ( $\epsilon$  phase, FCC structure) can be prepared by reacting manganese amidinate precursors with NH<sub>3</sub>

Excellent step coverage  
holes with AR = 52:1

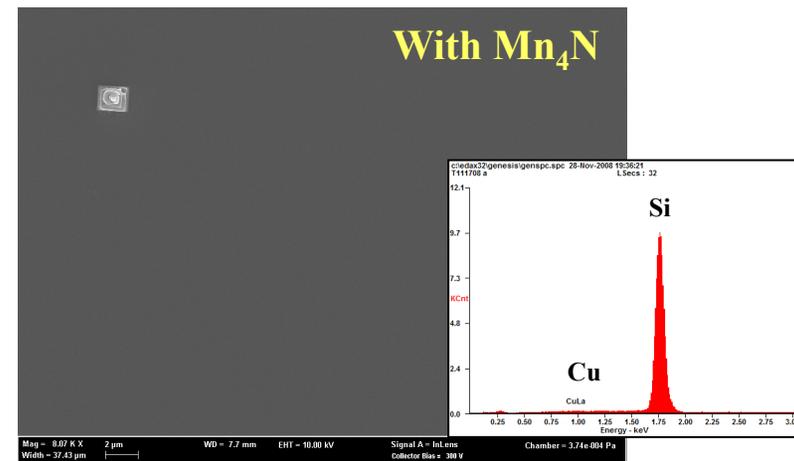
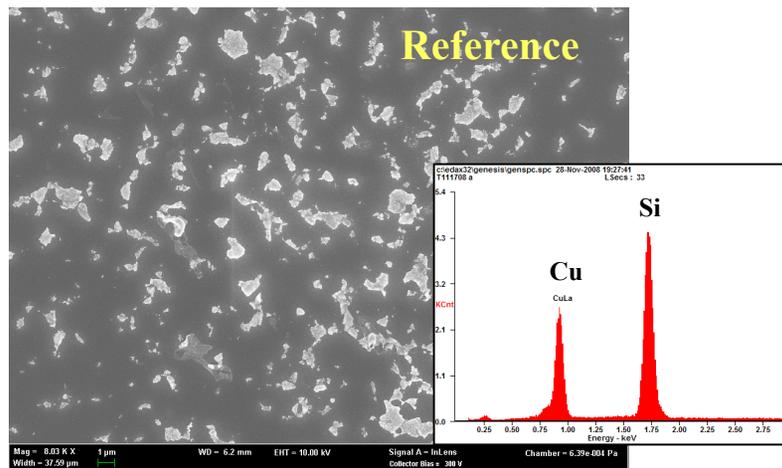
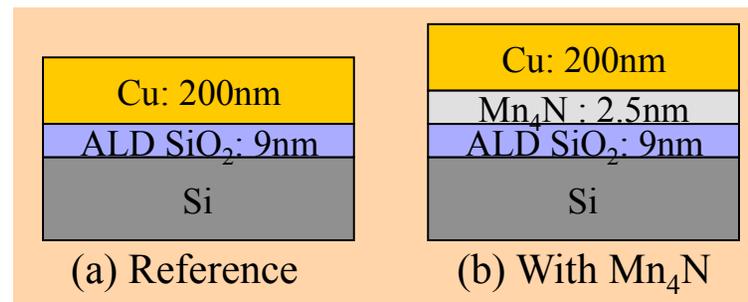


RMS roughness = 0.97 nm for a 13.5 nm film



# CVD of Manganese Nitride

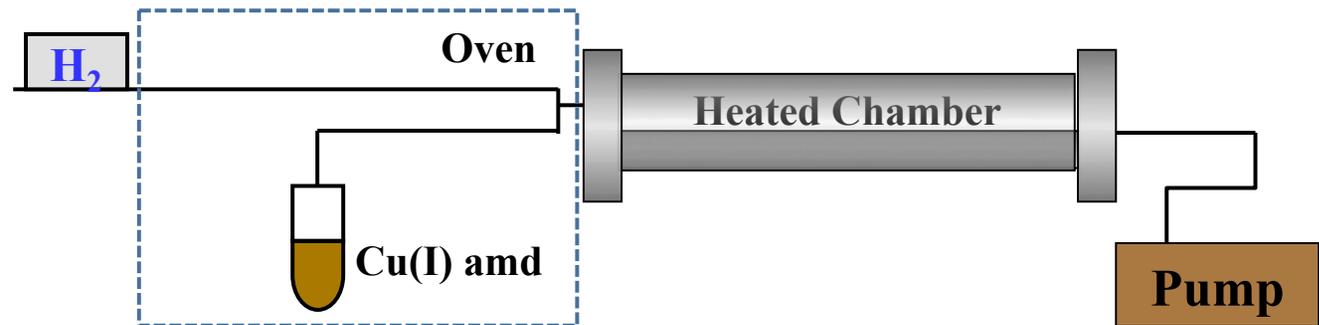
- $\text{Mn}_4\text{N}$  layer as thin as 2.5 nm can significantly improve adhesion between Cu and  $\text{SiO}_2$
- Four point bend: debonding energy =  $6.5 \text{ J/m}^2$  (D.E.  $\geq 5 \text{ J/m}^2$  to survive CMP)
- Thin  $\text{Mn}_4\text{N}$  layer also shows barrier properties against Cu diffusion



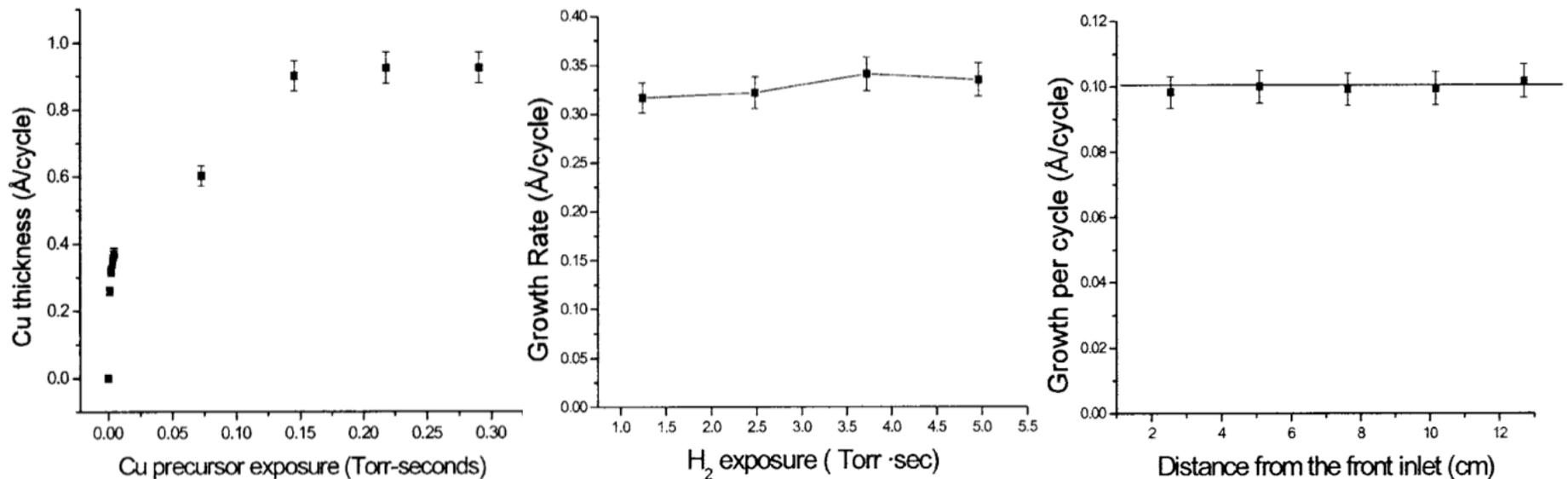
# ALD of Copper

- Copper films could be deposited by ALD using molecular hydrogen as reducing agent

## ALD System



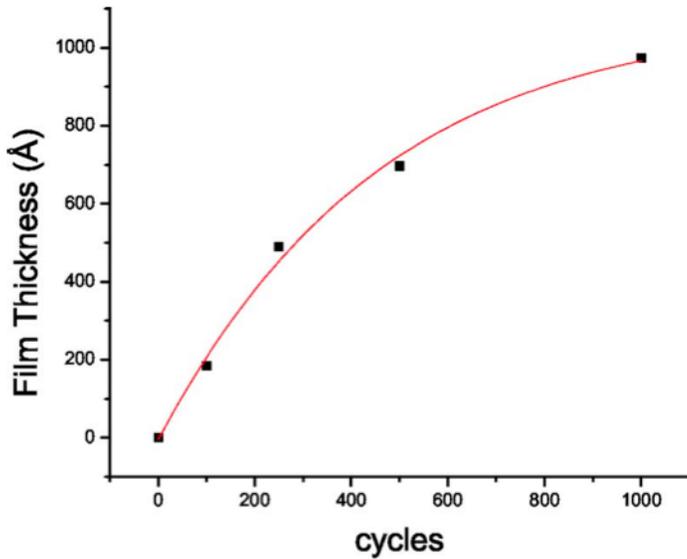
- Copper deposited on ALD- $Al_2O_3$  substrate at low temperatures (150-190°C):



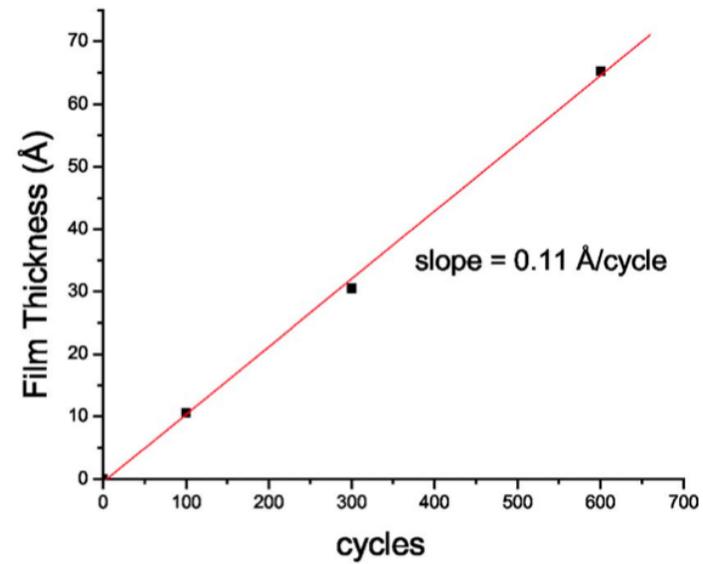
- Slow surface reaction suggests the possibility of highly conformal CVD reactions

# ALD of Copper

- Growth behavior can be affected by many factors: surface chemistry, precursor exposure, deposition temperature, etc.

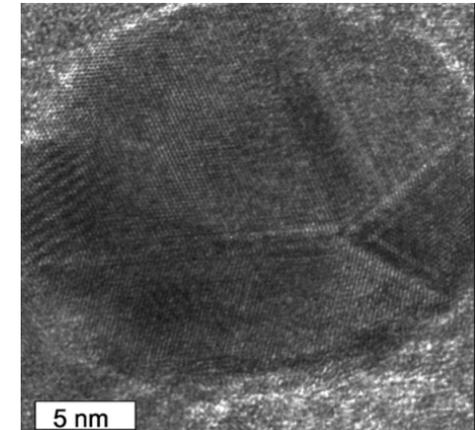
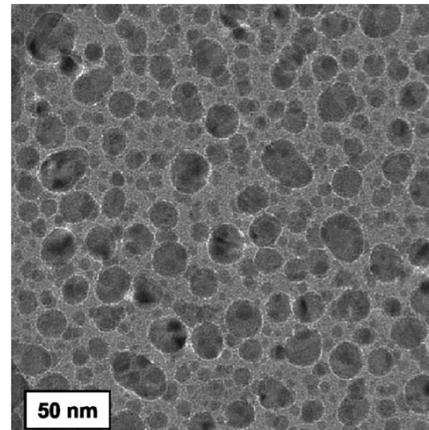


ALD-Al<sub>2</sub>O<sub>3</sub>, ALD-HfO<sub>2</sub>, Thermal SiO<sub>2</sub>  
Initially ~2Å/cycle, ~0.5Å/cycle when surface is fully covered by Cu



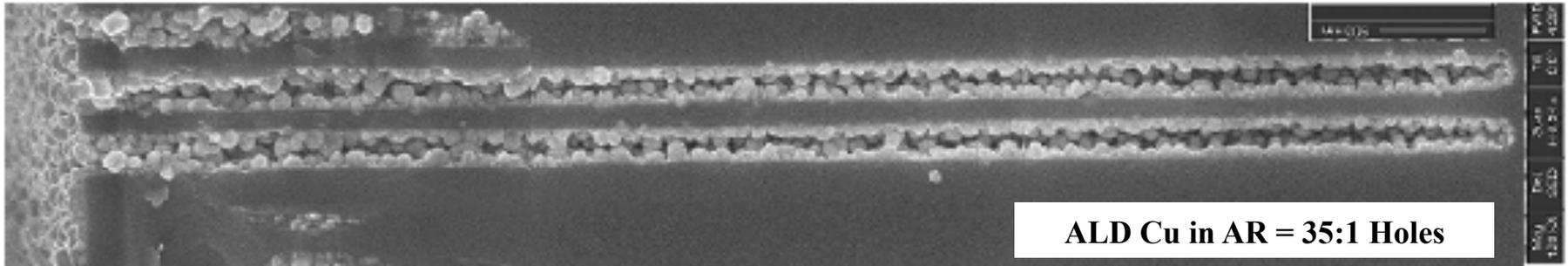
Ru Substrates  
0.11Å/cycle

Substrate	Growth per cycle (Å/cycle)
Al <sub>2</sub> O <sub>3</sub> /SiO <sub>2</sub>	1.90 (based on 100 cycles)
Si <sub>3</sub> N <sub>4</sub>	1.50 (based on 60 cycles)
WN	0.54 (based on 30 cycles)
Ru	0.11 (based on 100 cycles)
Co	0.40 (based on 30 cycles)
Cu	~0.5 (from Al <sub>2</sub> O <sub>3</sub> curve)



# ALD of Copper

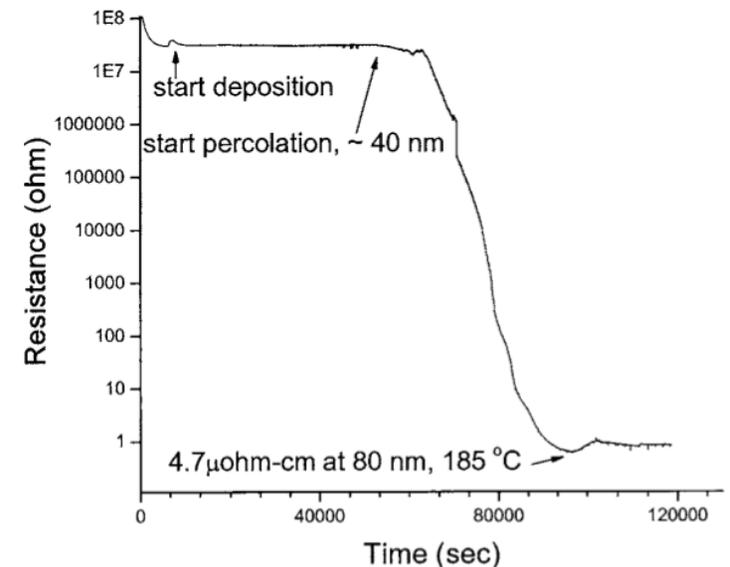
- ALD has the ability to grow films conformally and uniformly over high aspect ratio holes and trenches



- Four-point bend experiment showed high adhesion energies for Cu/Co/WN/SiO<sub>2</sub>

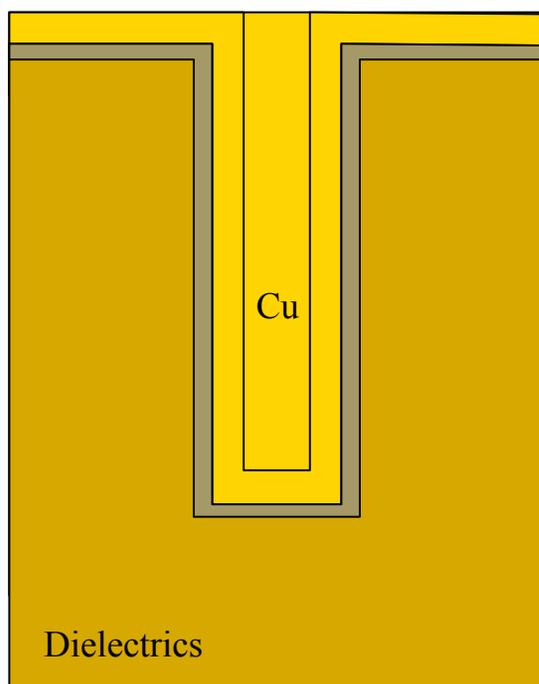
Structure	Scotch tape test	Adhesion energy (J/m <sup>2</sup> )
Co/SiO <sub>2</sub>	Failed	
Cu/SiO <sub>2</sub>	Failed	2 <sup>a</sup>
Cu/WN/SiO <sub>2</sub>	Failed	
TaN/SiO <sub>2</sub>	Passed	6 <sup>a</sup>
WN/SiO <sub>2</sub>	Passed	>31
Co/WN/SiO <sub>2</sub>	Passed	>31
Cu/Co/WN/SiO <sub>2</sub>	Passed	>31

## *In-situ* Resistance Measurement ALD Cu on Glass (185°C)



# CVD of Copper Manganese Alloy Seed Layer

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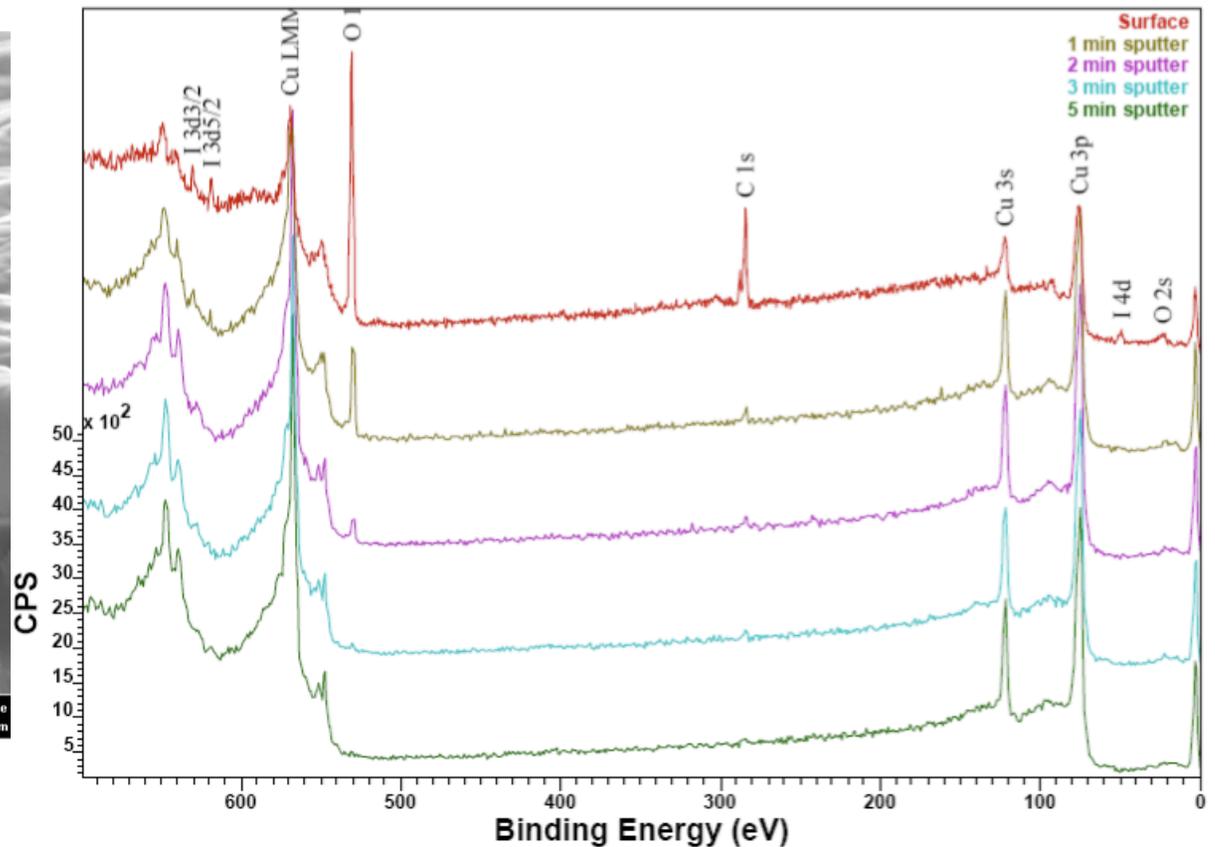
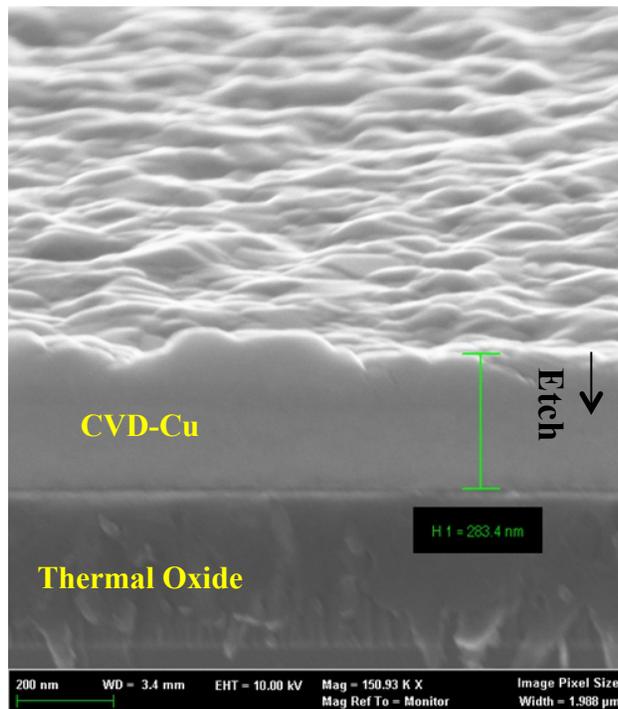
- Conformally deposited silica insulating layer plugs pores and makes clean SiO<sub>2</sub> surface
- Conformally deposited manganese nitride serves as a barrier/adhesion layer
- Iodine acts as a surfactant catalyst to increase Cu growth rate and make a smooth surface
- Conformally deposited copper-manganese alloy seed layer on manganese nitride
- Mn diffuses out from Cu during post-annealing to further improve adhesion and barrier properties at Cu/insulator interface

Y. Au, Y. Lin, H. Kim, E. Beh, Y. Liu and R. G. Gordon, *J. Electrochem. Soc.*, **157** (6) D341-D345 (2010).

Y. Au, Y. Lin and R. G. Gordon, *J. Electrochem. Soc.*, **158** (5) D248-D253 (2011).

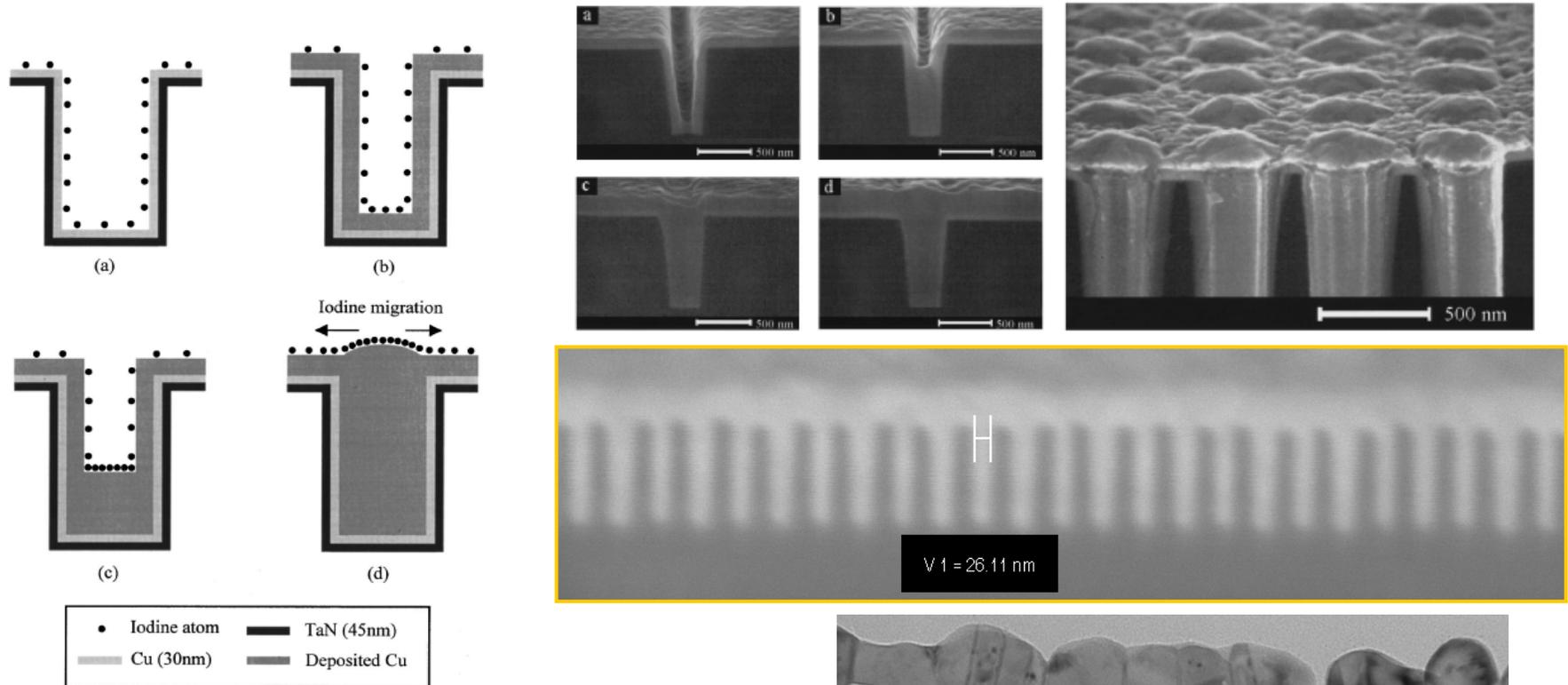
# Iodine Catalytic Surfactant

- Presence of iodine surfactant catalyst promotes higher (10x) deposition rate and smoother surface morphology
- Iodine is not incorporated into the film



# Void-free Bottom-up Filling of CVD-Cu

- Bottom-up filling of nanoscale features could be achieved with iodine catalyst



E. S. Hwang and J. Lee, *Chem. Mater.*, **12**, 2076 (2000).  
 K. Shim, O. Kwon, H. Park, W. Koh, and S. Kang, *J. Electrochem. Soc.*, **149** (2) G109-G113 (2002).  
 Y. Au, Y. Lin and R. G. Gordon, *J. Electrochem. Soc.*, **158** (5) D248-D253 (2011).

# Formation of Cu Seed Layer in High Aspect Ratio TSVs

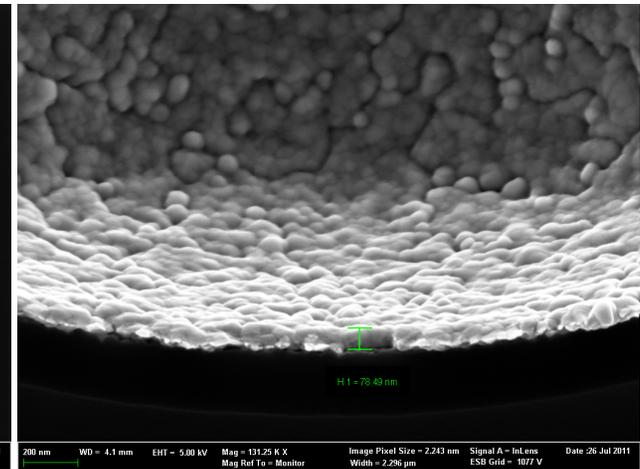
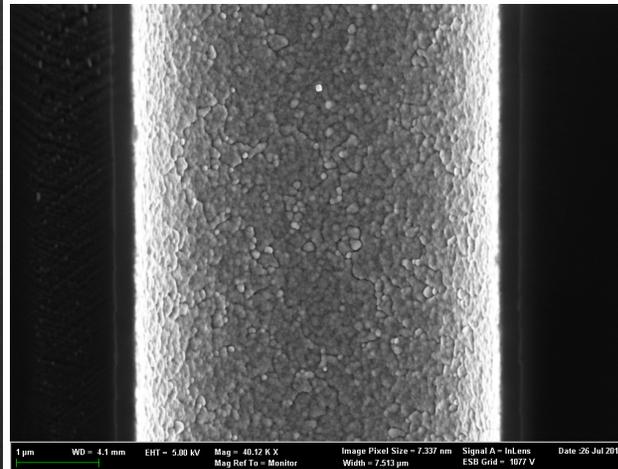
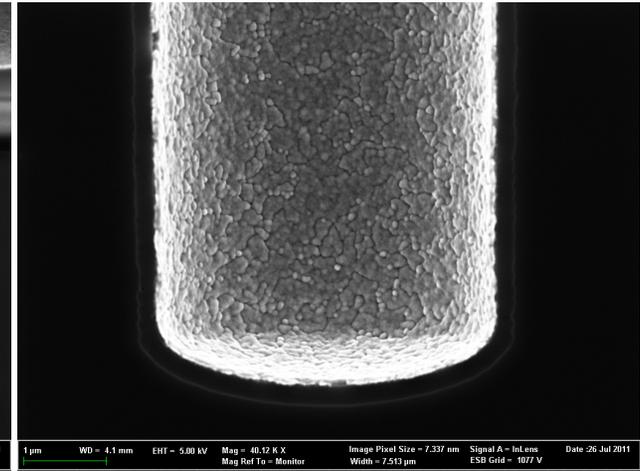
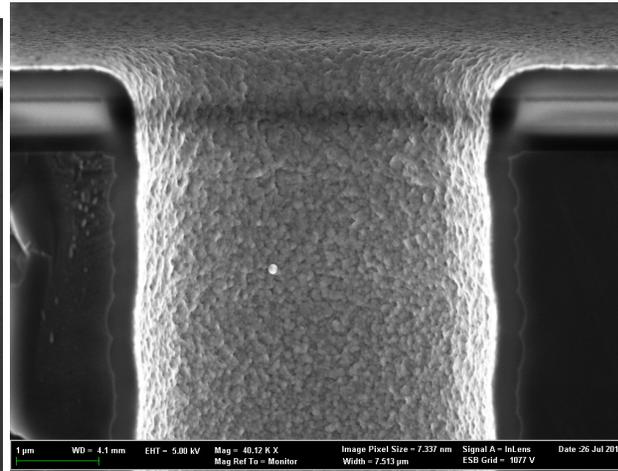
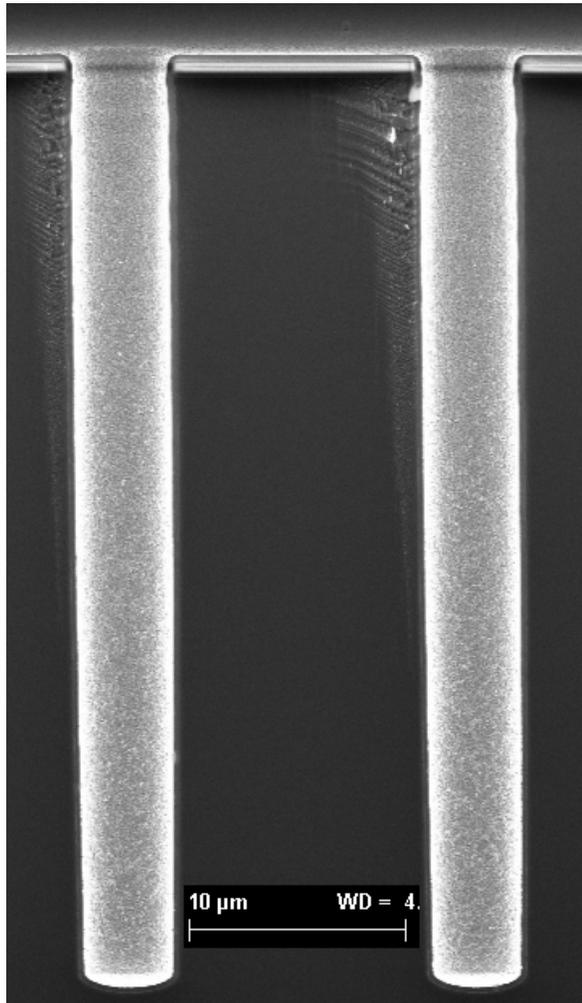
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CVD-Mn<sub>x</sub>N  
(130°C)

Iodine Exposure  
(Room Temp.)

CVD-CuMn  
(180°C)

Post-annealing  
(350°C)



Aspect Ratio = 10:1 TSV (IMEC)

- 70-80 nm of seed layer deposited conformally (~1 atomic % Mn)
- Sheet Resistance = 350-380 mΩ/sq.
- Strong adhesion

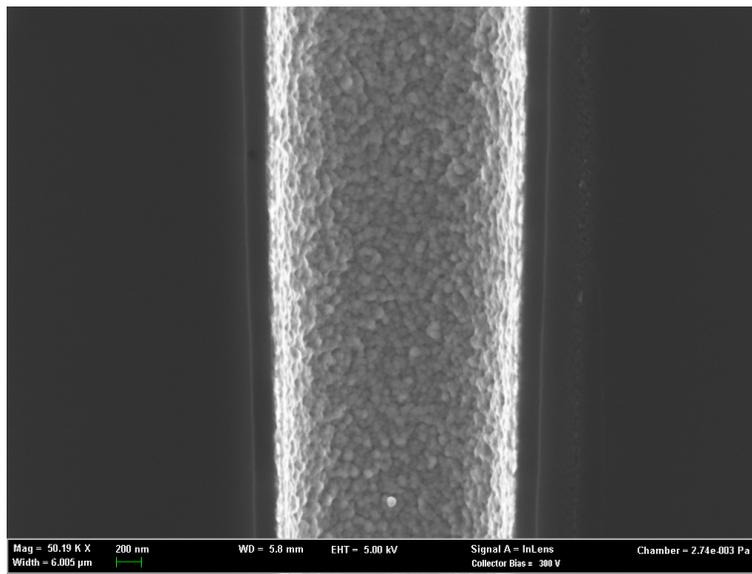
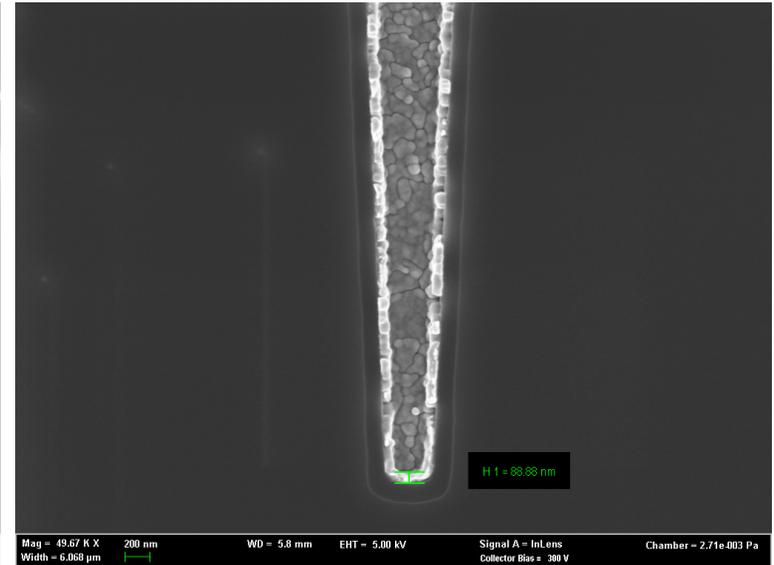
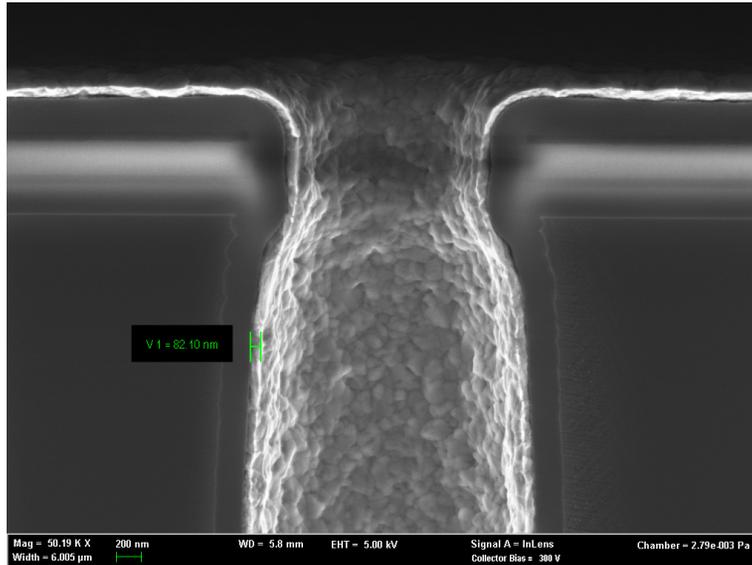
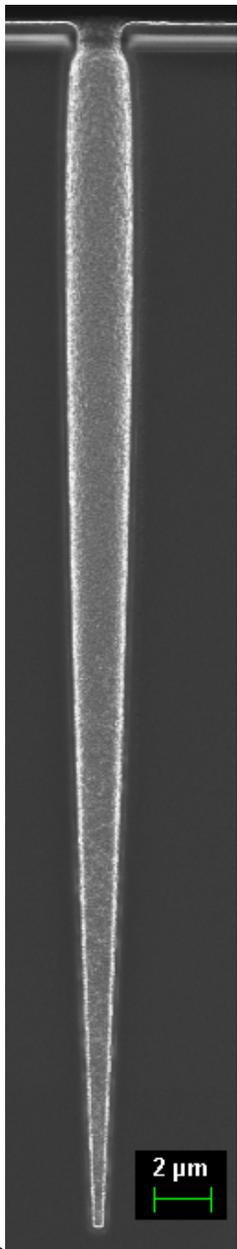
# Formation of Cu Seed Layer in High Aspect Ratio TSVs

CVD-Mn<sub>x</sub>N  
(130°C)

Iodine Exposure  
(Room Temp.)

CVD-CuMn  
(180°C)

Post-annealing  
(350°C)



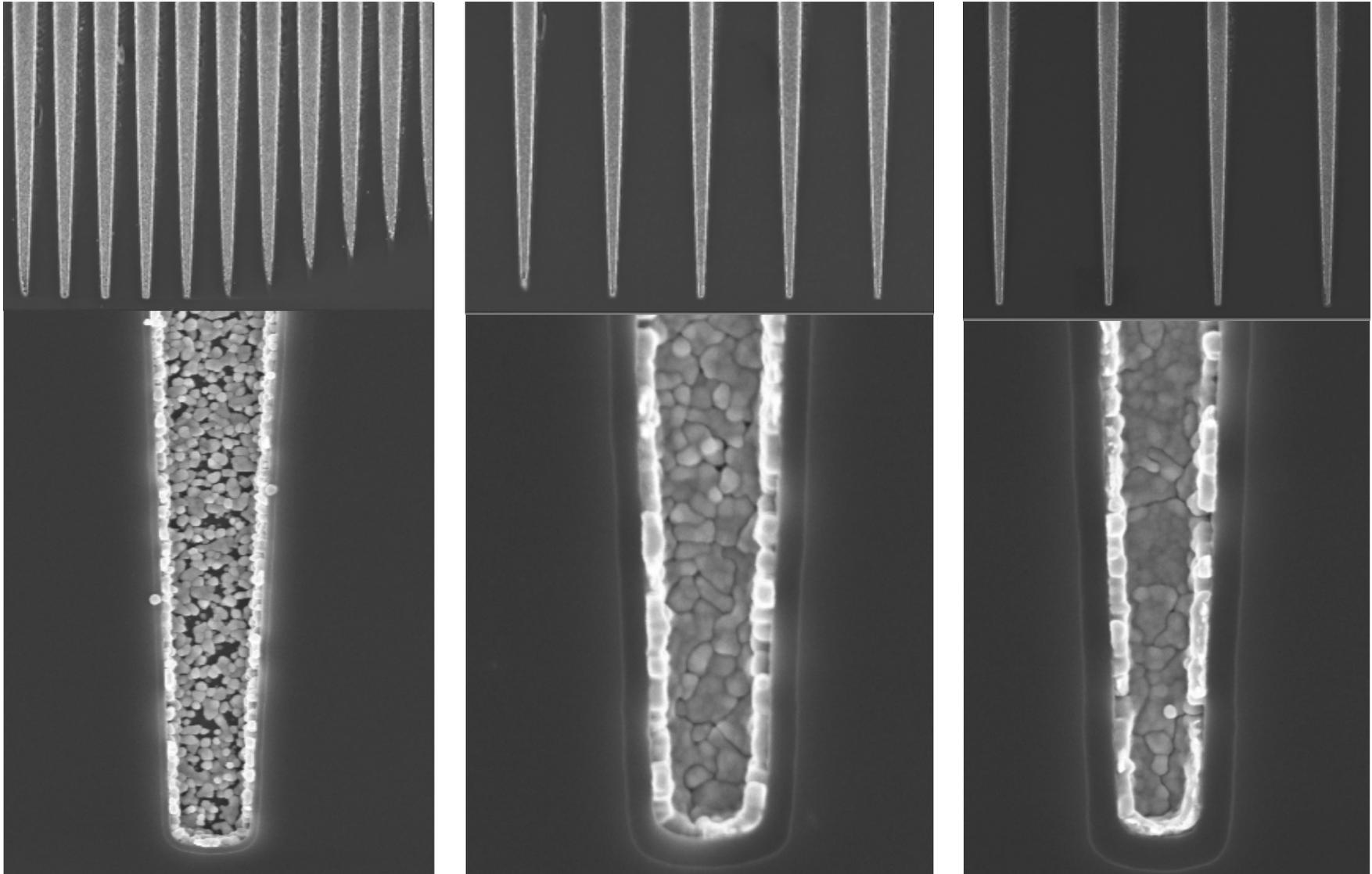
Aspect Ratio = 20:1 TSV (IMEC)

**70-80 nm of Seed Layer is Conformally Deposited**  
Sheet Resistance = 350-380 mΩ/sq.

## Formation of Cu Seed Layer in High Aspect Ratio TSVs

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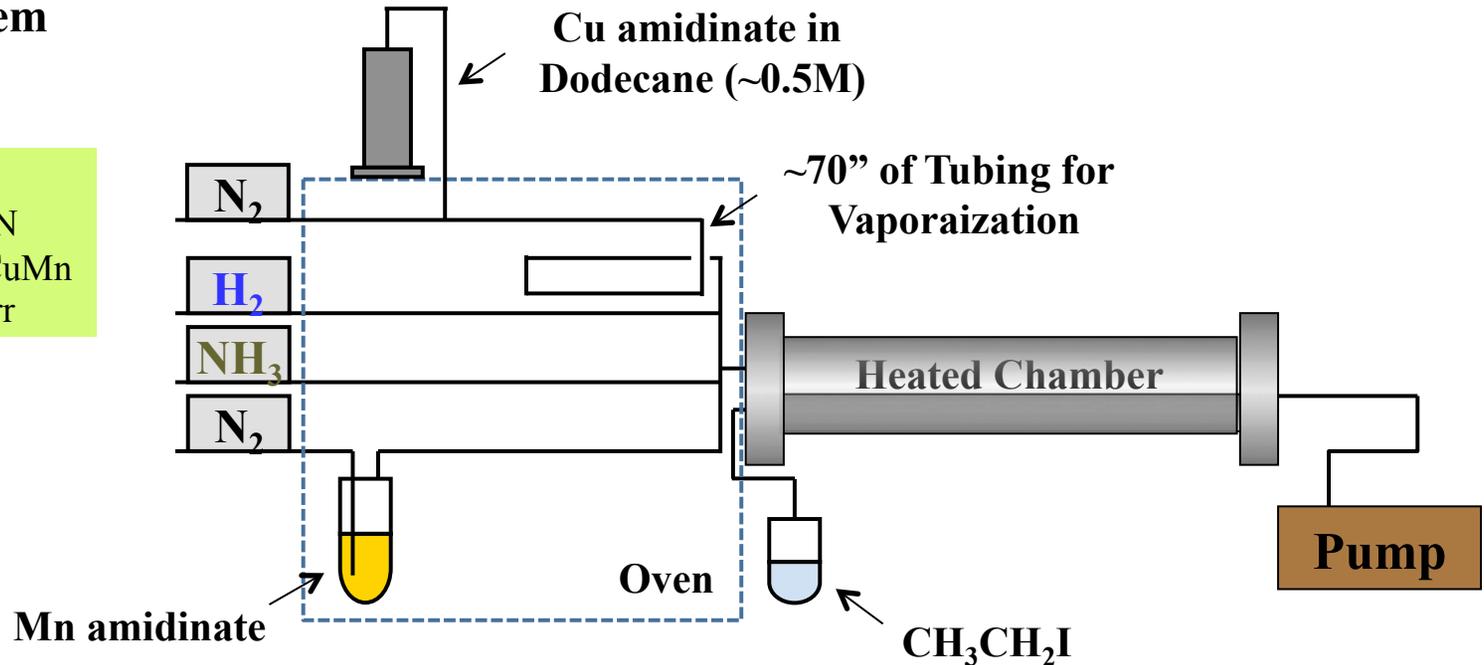
- Vapor delivery from a bubbler fails to nucleate densely at the bottoms of 20:1 TSV holes with extremely high density



# Direct Liquid Injection (DLI) CVD of Copper

## DLI-CVD System

**Temperature**  
 130°C for  $Mn_4N$   
 180°C for Cu and CuMn  
**Pressure: 5 Torr**



## Advantages of DLI-CVD Systems:

- High growth rate
- Predictable and consistent vapor delivery
- Good control of chemical composition of films
- Possibility to coat complex and porous substrates

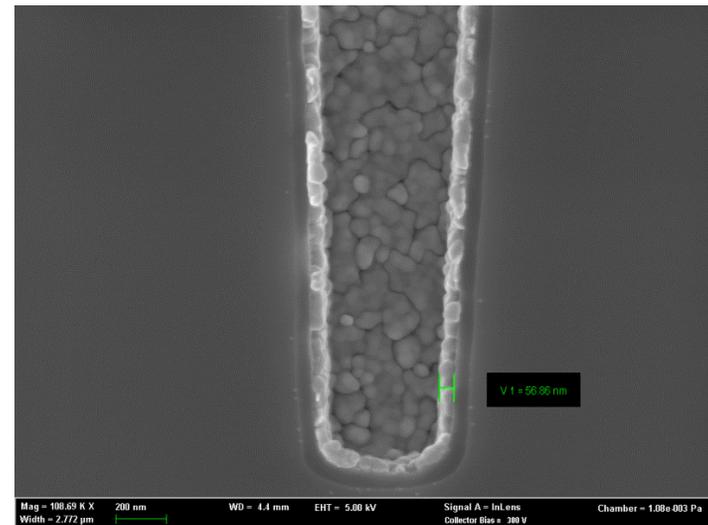
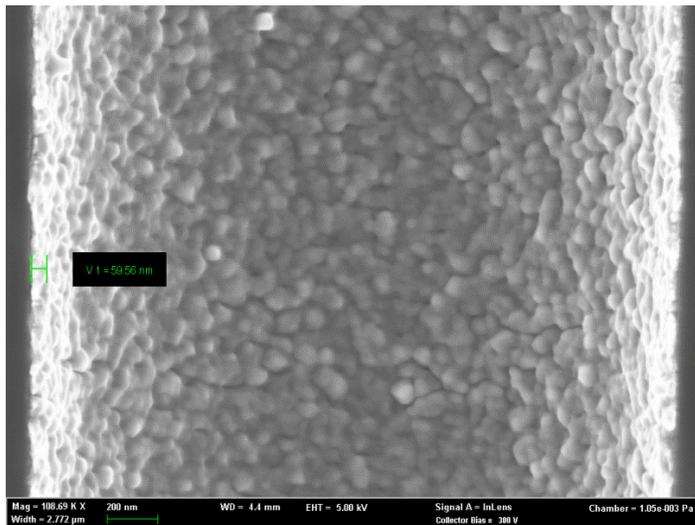
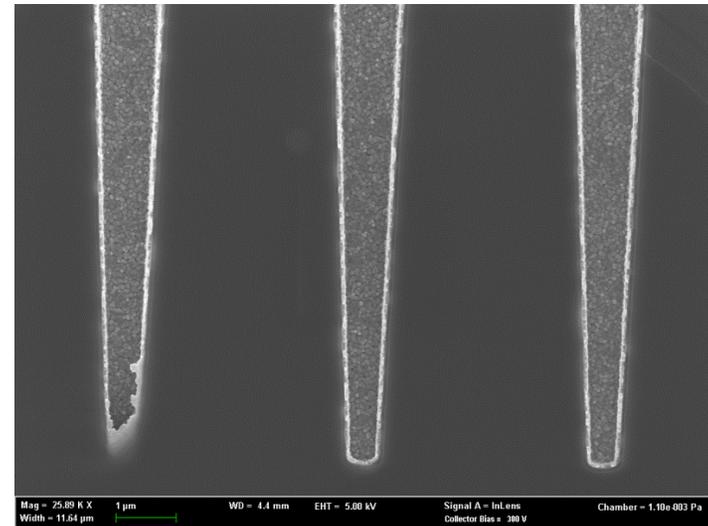
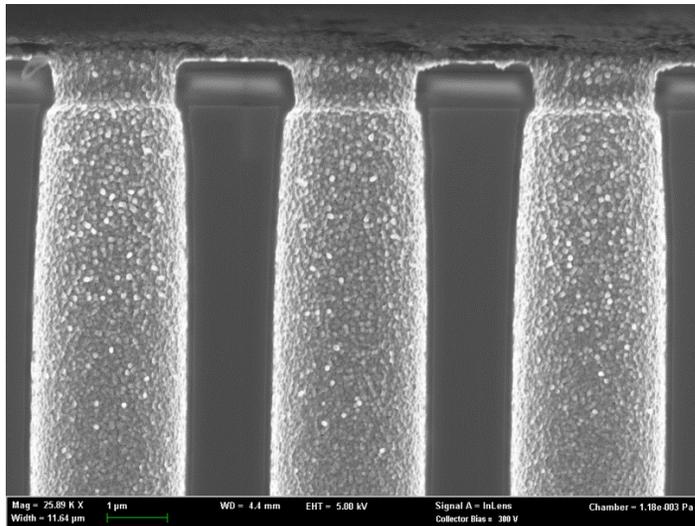
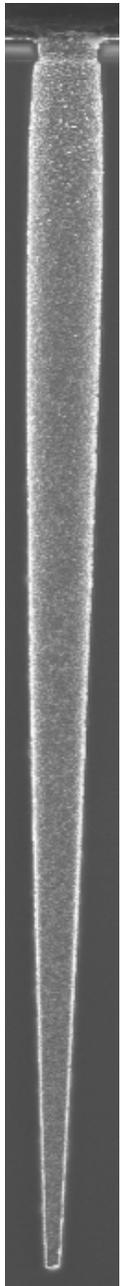
# DLI Cu-Mn Seed Layer in Dense 20:1 Aspect Ratio TSVs

CVD-Mn<sub>x</sub>N  
(130°C)

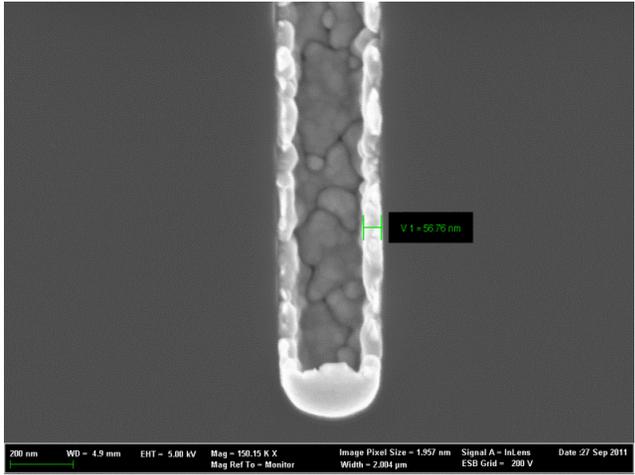
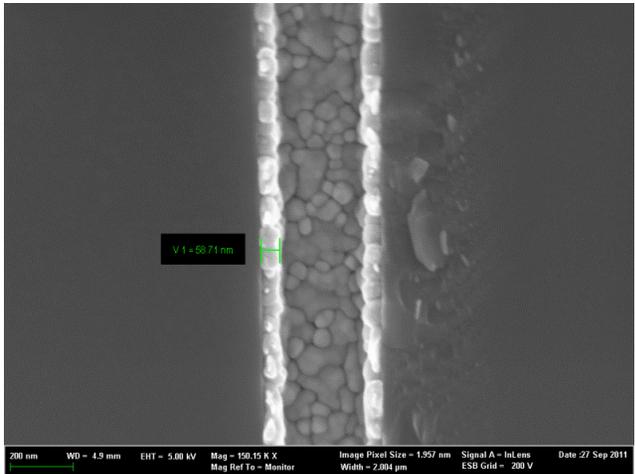
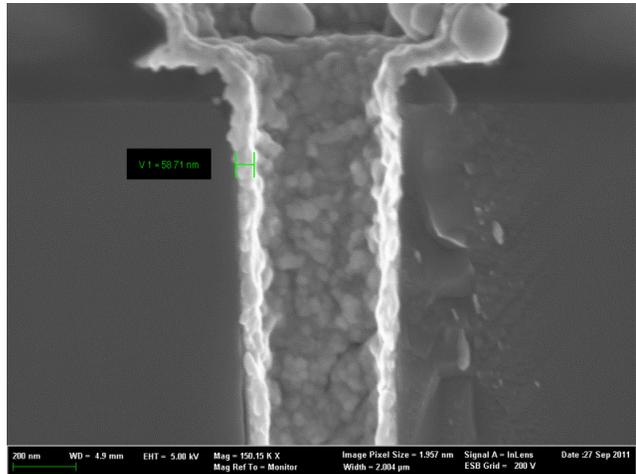
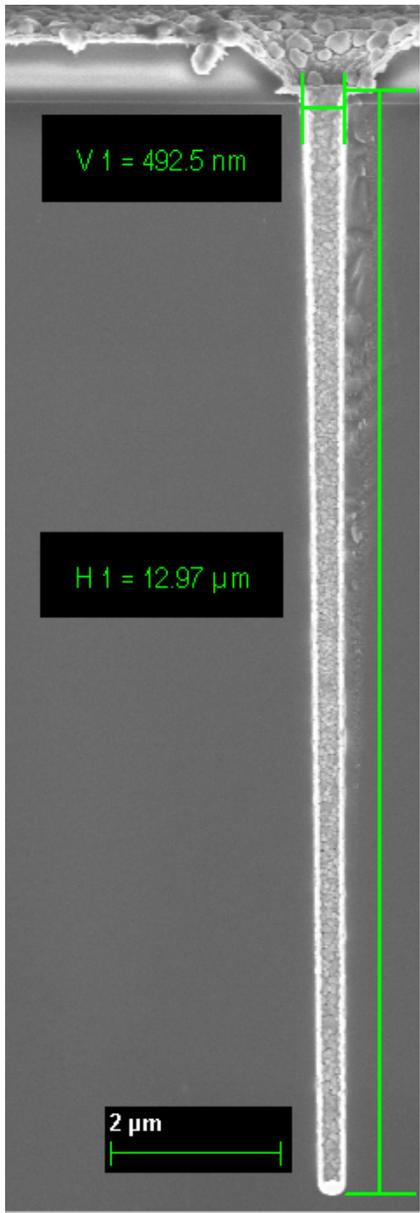
Iodine Exposure  
(Room Temp.)

CVD-CuMn  
(190°C)

Post-annealing  
(350°C)



# DLI Cu-Mn Seed Layer in Dense 25:1 Aspect Ratio TSVs

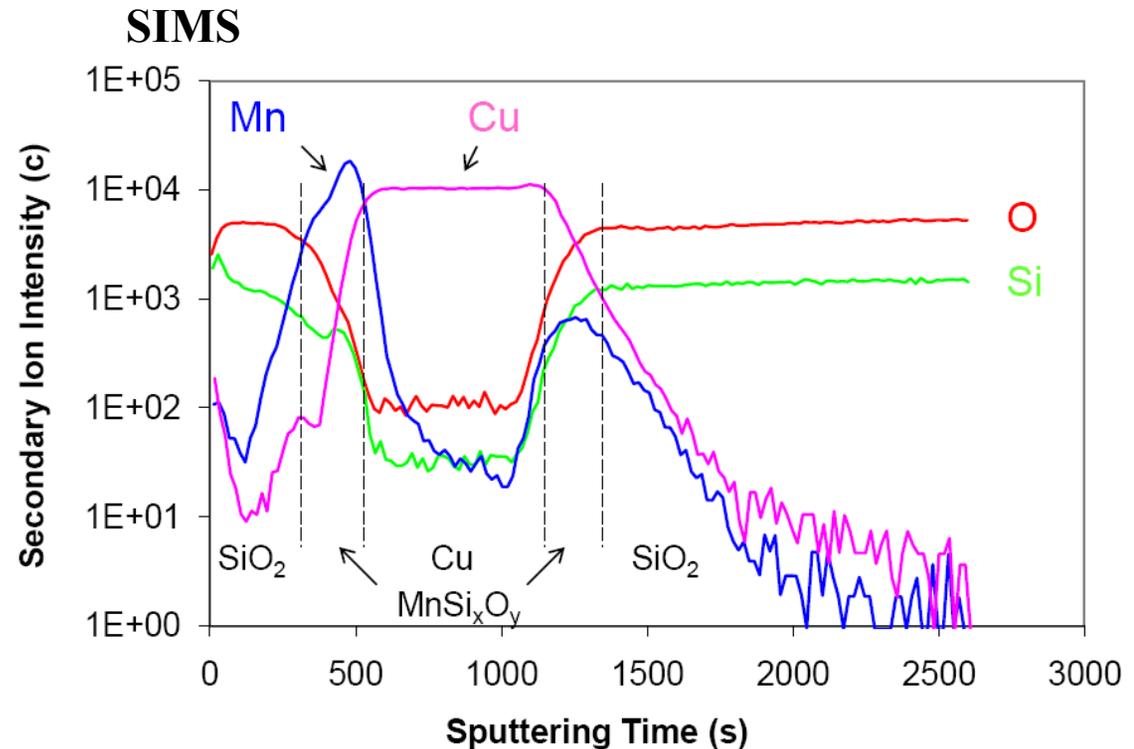
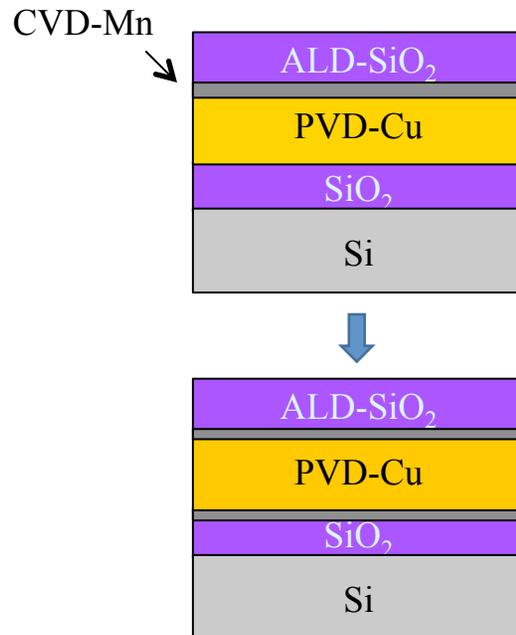


Aspect Ratio > 25:1 Holes

**TSV features with AR > 25:1 are conformally Coated by CuMn**

# Diffusion of Mn in Polycrystalline Cu

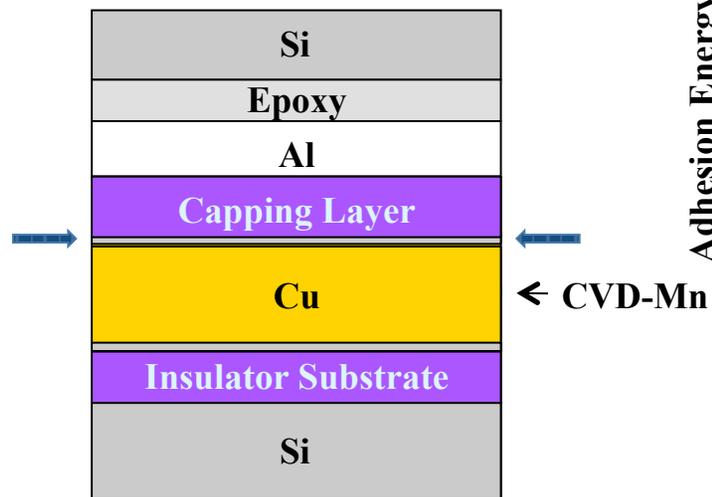
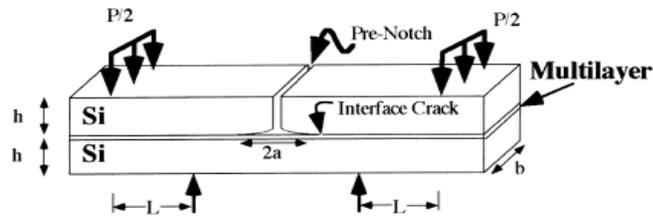
- Insulators ( $\text{SiO}_2$  and  $\text{Si}_3\text{N}_4$ ) encourage diffusion of Mn through Cu grain boundaries to Cu/ $\text{SiO}_2$  and Cu/ $\text{Si}_3\text{N}_4$  interfaces



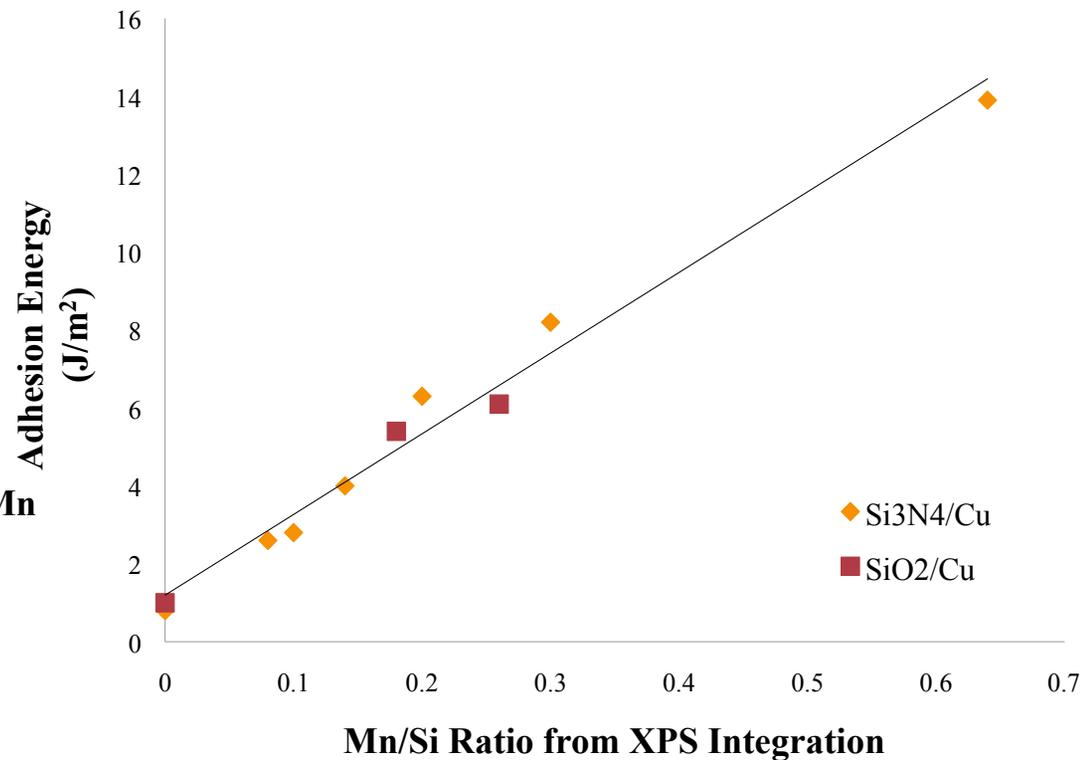
- XPS depth-profiling:  $D \approx 3 \times 10^{-21} \text{ m}^2/\text{s}$  at  $300^\circ\text{C}$
- CVD-CuMn process does not increase the resistivity of copper

# Adhesion Enhancement at Cu/Insulator Interface

- Four-point bend technique evaluates adhesion enhancement at Cu/insulator interfaces



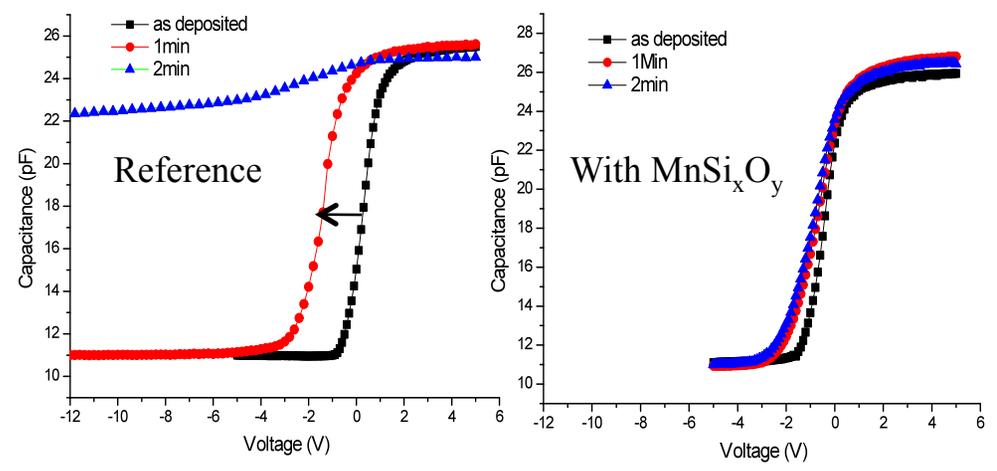
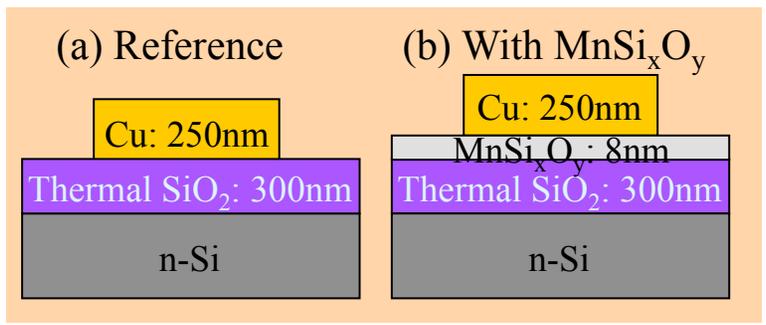
Adhesion Test Structure



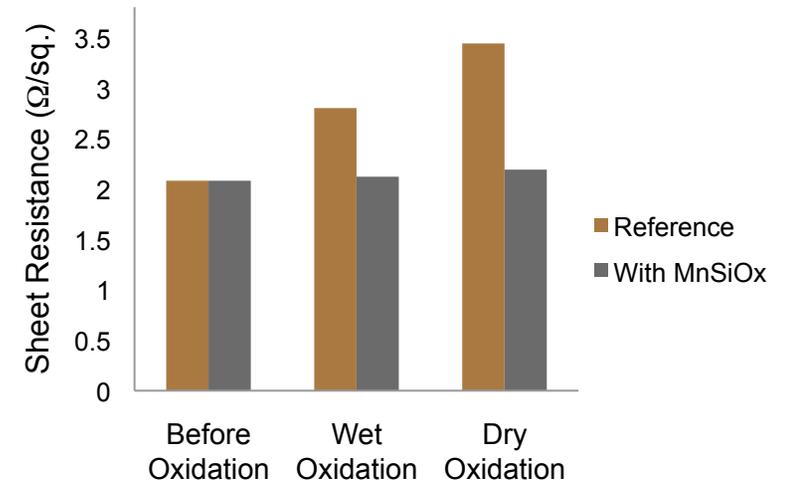
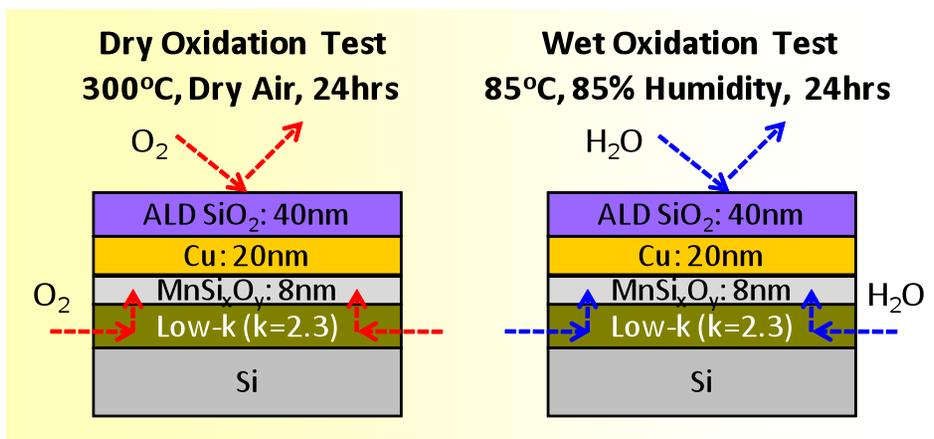
**Debonding energy at Cu/dielectric interface increases linearly with the amount of manganese at the interface**

# Manganese Silicate Diffusion Barrier

## Cu Diffusion Barrier Test



## Oxidation Barrier Test

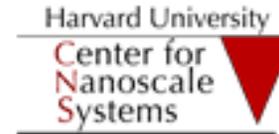


**MnSi<sub>x</sub>O<sub>y</sub> layer formed at the interface is an excellent barrier against diffusion of Cu, H<sub>2</sub>O and O<sub>2</sub>**

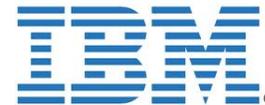
- ✓ Conformal ALD-silica can be deposited using TMA and silanol
- ✓ Conformal CVD-manganese nitride barrier/adhesion layer can be prepared by reacting amidinate precursors with ammonia
- ✓ TSV features with AR > 25:1 can be conformally coated by CuMn seed layer using a direct liquid injection (DLI) CVD method
- ✓ Manganese in CuMn alloy diffuses out to strengthen the interface between Cu and insulators without increasing the resistivity of Cu
- ✓ Manganese silicate ( $\text{MnSi}_x\text{O}_y$ ) interfacial layer shows excellent barrier properties against Cu diffusion and protects Cu from corrosion by  $\text{H}_2\text{O}$  and  $\text{O}_2$

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- ALD of Cu: Antti Rahtu, Zhengwen Li
- CVD of Cu and Mn: Hoon Kim
- CVD of Manganese Nitride and Adhesion: Youbo Lin

